

## MORE QUESTIONS AND ANSWERS

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### Is $\text{Mg}_3\text{N}_2$ ionic or molecular?

Answer and Explanation:  $\text{Mg}_3\text{N}_2$  is an ionic compound. Electronegativity is a measure of how strongly an atom attracts electrons in a chemical bond.

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### RELATED QUESTIONS

### Is $\text{Cr}_2\text{O}_3$ molecular or ionic?

Chromium trioxide ( $\text{Cr}_2\text{O}_3$ ) is an ionic compound composed of 2  $\text{Cr}^{3+}$  ions, and 3  $(\text{O})^{2-}$  ions. (That's an oxygen ion with a -2 charge). In the formation of this compound electrons are not shared between the constituent ions, as they would be between atoms in a covalent bond.

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### Is $\text{BaO}$ a molecular or ionic compound?

ionic crystals ionic binary crystals such as barium oxide ( $\text{BaO}$ ), calcium sulfide ( $\text{CaS}$ ), barium selenide ( $\text{BaSe}$ ), or strontium oxide ( $\text{SrO}$ ). They have the same structure as sodium chloride, with each atom having six neighbours. Oxygen can be combined with various cations to form a large number of ionically bonded solids.

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