MORE QUESTIONS AND ANSWERS

What is the charge of HClO3?

It can hold more than 8 valence electrons. So we really should check our formal charges to see if this is the best structure for HClO3. When we calculate the formal charges, we find that the Chlorine has a +2 charge, while these Oxygens out here have a -1 charge, each of them.

RELATED QUESTIONS

What is the oxidation state of CI in HCIO3 answers com?

oxidation number of cl ls +5.

What is the oxidation number of Na of CL?

The oxidation number of sodium in the Na+ ion is +1, for example, and the oxidation number of chlorine in the Cl- ion is -1.



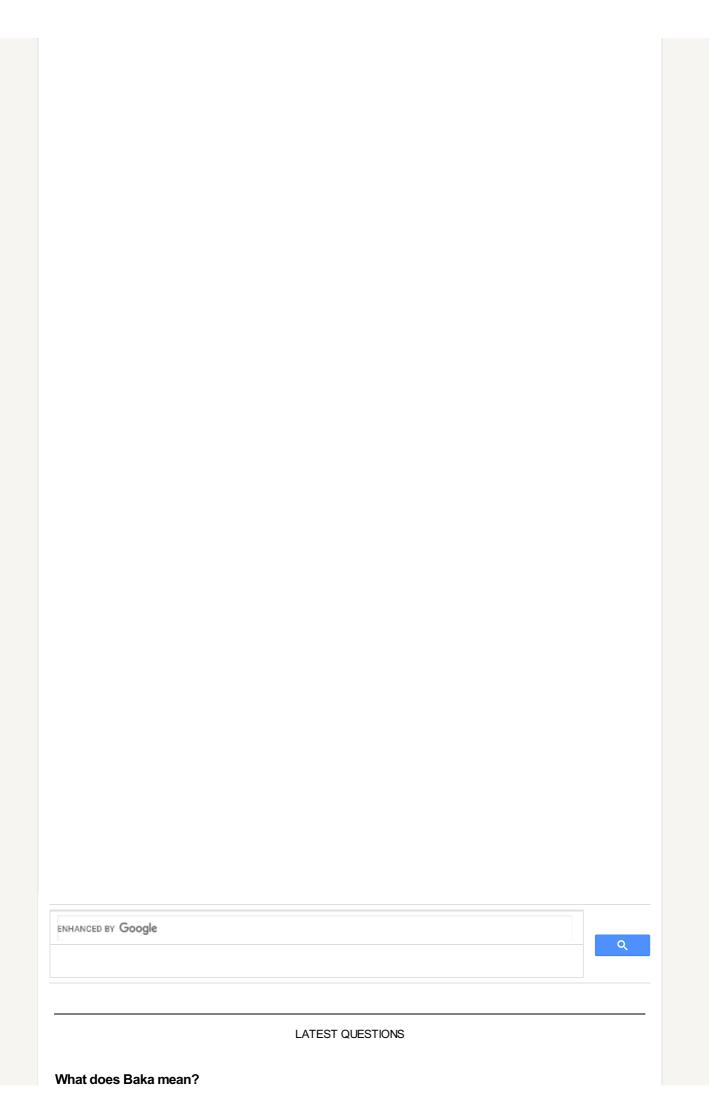












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