MORE QUESTIONS AND ANSWERS

What is the oxidation number of Na of CL?

The oxidation number of sodium in the Na+ ion is +1, for example, and the oxidation number of chlorine in the CI- ion is -1

What is oxidation number of Na CI?

CI has an oxidation number of -1, as the sum of the oxidation numbers is zero the oxidation number of S = +1 What are the oxidation numbers in the compound NaCl? To Na nox is +1 and for CI is -1.

What is the oxidation number of CI in NaOCI?

+1 Firstly it's an ionic compound so Na+ and OCI-. In OCI- Oxygen is more electronegative than Chlorine hence its oxidation state remains constant at -2 leftover is Chlorine to balance the charges on OCI- ot has to be of the oxidation state +1

What's the oxidation number of sodium chloride?

Here is a really nice answer on oxidation numbers from Khan Academy; Oxidation number. So the oxidation number of the sodium ion is +1 and the chloride ion is -1. Together Sodium chloride has an oxidation number of 0.

RELATED QUESTIONS

What is the oxidation state of CI in HCIO3 answers com?

oxidation number of cl ls +5.

What is the charge of HCIO3?

It can hold more than 8 valence electrons. So we really should check our formal charges to see if this is the best structure for HClO3. When we calculate the formal charges, we find that the Chlorine has a +2 charge, while these Oxygens out here have a -1 charge, each of them.







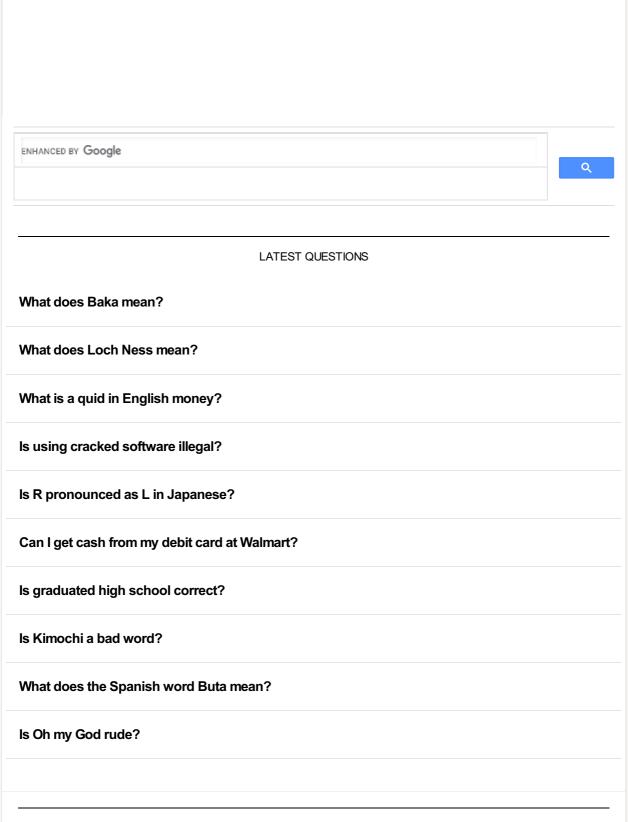












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