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What is the oxidation number of Na of CL?

The oxidation number of sodium in the Na^+ ion is +1, for example, and the oxidation number of chlorine in the Cl^- ion is -1.

What is oxidation number of Na Cl?

Cl has an oxidation number of -1, as the sum of the oxidation numbers is zero the oxidation number of S = +1 What are the oxidation numbers in the compound NaCl? To Na nox is +1 and for Cl is -1.

What is the oxidation number of Cl in NaOCl?

+1 Firstly it's an ionic compound so Na^+ and OCl^- . In OCl^- Oxygen is more electronegative than Chlorine hence its oxidation state remains constant at -2 leftover is Chlorine to balance the charges on OCl^- it has to be of the oxidation state +1

What's the oxidation number of sodium chloride?

Here is a really nice answer on oxidation numbers from Khan Academy; Oxidation number. So the oxidation number of the sodium ion is +1 and the chloride ion is -1. Together Sodium chloride has an oxidation number of 0.

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oxidation number of Cl is +5.

What is the charge of HClO3?

It can hold more than 8 valence electrons. So we really should check our formal charges to see if this is the best structure for HClO_3 . When we calculate the formal charges, we find that the Chlorine has a +2 charge, while these Oxygens out here have a -1 charge, each of them.

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