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What is the oxidation state of Cl in HClO₃ answers com?

oxidation number of Cl is +5.

What is the oxidation state of S in HSO₄²⁻?

That depends on the compound that it is in but it is often found in the oxidation states 2-, 4+, and 6+. In its elemental form the oxidation state is 0, which is true of all elements.

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What is the oxidation number of Na of Cl?

The oxidation number of sodium in the Na⁺ ion is +1, for example, and the oxidation number of chlorine in the Cl⁻ ion is -1.

What is the charge of HClO₃?

It can hold more than 8 valence electrons. So we really should check our formal charges to see if this is the best structure for HClO₃. When we calculate the formal charges, we find that the Chlorine has a +2 charge, while these Oxygens out here have a -1 charge, each of them.

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