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### Which substance contains a multiple covalent bond?

Carbon dioxide is another example of a compound containing a multiple bond. Carbon can react with oxygen to form carbon dioxide. Carbon has four valence electrons, and oxygen has six. Carbon can share two of its valence electrons with each of the two oxygen atoms, forming two double bonds.

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#### What are covalent molecules composed of?

A covalent bond is formed by two atoms sharing a pair of electrons. The atoms are held together because the electron pair is attracted by both of the nuclei. In the formation of a simple or ordinary covalent bond, each atom supplies one electron to the bond – but that does not have to be the case.

#### Which elements are made up of molecules?

2.8. Elements can be made of one atom, like He, or be elemental molecules, such as hydrogen (H<sub>2</sub>), oxygen (O<sub>2</sub>), chlorine (Cl<sub>2</sub>), ozone (O<sub>3</sub>), and sulfur (S<sub>8</sub>).

#### Which molecule contains a total of three covalent bonds?

A triple covalent bond is a covalent bond formed by atoms that share three pairs of electrons. The element nitrogen is a gas that composes the majority of Earth's atmosphere. A nitrogen atom has five valence electrons, which can be shown as one pair and three single electrons.





























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